Tetrahedron Letters No.36, pp. 2443-2450, 1964. Pergamon Press Ltd. Printed in Great Britain.

AN APPROACH TO THE SYNTHESIS OF IBOGAINE

Stephen I. Sallay

Research Division, Wyeth Laboratories Radnor, Pennsylvania (Received 22 June 1964; in revised form 10 July 1964)

IN view of the increasing interest in iboga alkaloids I should like to present some of the preliminary steps toward a total synthesis of ibogaine (I) (1). The successful, stereochemically controlled synthesis of the tetracyclic indole (XVIa) suggests a feasible pathway for the synthesis of I and some of its congeners. The cis-fused \mathcal{G} rings of XVIa were constructed from the cis-enedione (IIa) (2), which, after suitable modifications (IIa \longrightarrow IIIa \longrightarrow VIa), was subjected to the Beckmann rearrangement to give VIIIa. The lactam was then reduced to the cis-aminoketal (Xa). The aminoketone derived from Xa underwent indole formation, producing the A-D ring system of XVIa and ibogaine (I).

Parallel transformations of the stereochemically more stable transenedione (IIb) were carried out, the availability of the trans series (IIIb-VIIIb and Xb) being helpful in determining the configurations during every step of the synthesis. Gas chromatography proved that the separately equilibrated IIa and IIb isomers reached a cis/trans ratio of 1:5.7. In spite of the stereochemical instability of IIa $(3, 4)$, it proved to be a useful starting material. The isolated double bond of both epimers (IIa,b) survived all the steps and was found to be particularly helpful in verifying the structures of VIa, b and VIIIa, b (vide infra).

2443

In reproducing the preparation of the trans-enedione (IIb), as reported by Henbes:, et al. (5), it was proved that the intermolecular chelate (XIV) $[m, p, 159.160^{\circ}; \lambda_{\text{max}}^{K} 215, 292 \text{ m} (67100, 3000); \lambda_{\text{max}}^{KBr} 3.08 (OH), 5.92 \mu$ $(SC=0)$] was the actual product isolated. The two components of XIV were separated either by thin-layer chromatography (T.L.C.) (R_f 0.33 and 0.38)^a or (in 90%, yield) by solvolysis. The more puckered cis-dione (IIa) did not chelate with the planar hydroquinone derivative (XV) (6). This difference in behavicr offered a means for the isolation of the trans-enedione (IIb) from an equilibrated epimer mixture in 75% over-all yield. The m.p. of IIb was found to be identical (95.5-96.5') with that reported by Ireland and Marshall (3). The conspicuous differences between the nuclear magnetic resonance $(n,m,r.)$ spectra^b of the two epimers (IIa, b) established their conformations. The spectrum of IIa indicated non-equivalence for its four C_2 , C_3 -protons (sharp peaks at δ 2.77 and 2.80 p.p.m.). Contrary to this observation, equivalence of the protons in similar positions of IIb $(6, 2.72 \text{ p.p.m.}, s)$, analogously to zyclohexane-1, 4-dione (7-9), suggested a twisted-boat conformation for its A ring. The axial-equatorial C_9 , C_{10} -protons of IIa were easily distinguishable (δ 3.17 p.p.m., m) from the similar but diamagnetically shifted axial-axial protons of IIb $(6, 2.59 \text{ p.p.m.}, \text{ m}).^C$

a) The T.L.C. systems were : Al_2O_3 -G[ethyl acetate - n-hexane (2:3)] for XIV; neutral silica gel — starch (10) [ethyl acetate — $_{\rm n}$ -hexane (2:3) for IIa, b to VIa, b and XXIIIa, b; Al₂O₃-G[chloroform] for VIIIa, b; $Al_2O_3-G[chloroform - cyclohexane - diethylamine (7:2:1)]$ for XVIa,b and XVII.

b) Measured in deuteriochloroform at 60 MC on a Varian, Model A-60, spectrometer and expressed as p.p.m. shift (δ) downfield from tetramethylsilane.

C) Details of the conformational analyses will be published elsewhere.

* The ring systems of series a and b possess cis and trans configurations, respectively.

The cis-dione (IIa) failed to produce a satisfactory yield of the <u>cis</u>-moncxime (Va) [$\chi^{KBT}_{\rm max}$ 3.15 (OH), 5.85 $_{\rm \mu}$ (∞ =0); R_f 0.17^a] and, furthermore Va could not be rearranged to the corresponding lactam. To achieve a higher degree of stercostability before the oxime function was introduced, the monoketalization of IIa was studied (3,4). The ketalization, which was followed by quantitative T.L.C., led to a mixture of the <u>cis</u>-monoketal (IIIa,yi ca. 27%) [m.p. $62-64^{\circ}$; $\lambda_{\text{max}}^{\text{KBr}}$ 5.85 μ ; 6 2.30 (8H,m), 3.09 (2H,m), 4.0 (4H, d), 5.52 p.p.m. $(2H, b)$; R_f 0.43] and the cis-bisketal (IVa, yield 10-15%) [m.p. $116-117$ ° (4); δ 1.77 (4H,m), 2.13 (6H,b), 3.86 (8H,s), 5.54 p.p.m. (2H,b); R_f 0.53], as well as to the trans-monoketal (IIIb, yield ca. 22%) [m.p. 52-54[°] (4); $\lambda_{\text{max}}^{\text{KB}}$: 5.85 μ ; 6 2.03 (8H,m), 2.56 (2H,m), 3.98 (4H,b), 5.53 p.p.m. (2H,b); R_f 0.48] and the trans-bisketal (IVb, yield 15-30%) [m.p. 97.5-98° (3); δ 1.71 $(4H, s)$, 2.01 (6H,s), 3.86 (8H,b), 5.52 p.p.m. (2H,b); R_f 0.55]. Some starting , material (IIq ca. 5%) (R_f 0.24) and its epimer (IIb, ca. 10%) (R_f 0.35) were also detected. Thus, the T.L.C. and n.m.r. data revealed the concomitant epimerization and two-step ketalization of IIa. It was also proved that the cis-monoketal (IIIa) did not equilibrate with the trans-epimer (IIIb) during isolatior from a Florisil column, as reported by others (4).

The two monoketals (IIIa,b) retained their configurations during oximation and gave rise to the <u>cis-anti</u>-oximeketal (VIa, yield 27% calculated on IIa) [m.p. 178-179°; λ_{mR}^{K} 3.15 (OH), 5.98 (∞ =N-), 6.03 μ (∞ =C<); R_f 0.46] and <u>trans-anti</u>-oximeketal (VIb, yield 22% calculated on IIa) [m.p. 166-167° $\lambda_{\text{max}}^{\text{KBr}}$ 3.15 (OH), 5.98 (∞ =N-), 6.03 u (∞ =C<); R_f 0.54]. The <u>anti</u>-oxime structure of VIa, b was verified by degradation (vide infra). On tosylation of the epimer ketaloximes (VIs,b) in warm pyridine, they exhibited a remarkable difference. While the trans-epimer (VIb) furnished the expected transtosyloximz ketsl (VIIb) (m.p. 131-132') in almost quantitative yield, the <u>cis</u>-epimer (VIa) spontaneously rearranged to the desired <u>cis</u>-lactamket (VIIIa, yield 89%) [m.p. 211.5-212^o; $\lambda_{\text{max}}^{\text{KBr}}$ 3.15 (NH), 6.03 u (∞ =0); R_f 0.34^a].

No.36 An approach to the synthesis of ibogaine 2447

It is believed that the coplanarity of the participating centers of the intermediate <u>cis</u>-tosyloxime (VIIa) facilitate the Beckmann rearrangement. The <u>trans</u>-tosyloxime (VIIb) was ring expanded to the trans-lactamketal (VIIIb) [m.p. 200°; $\lambda_{\rm max}^{\rm KBT}$ 3.15 (NH), 6.0 $_{\rm \mu}$ (>C=0); R_f 0.20] on a basic aluminum oxide column (11) in 78% yield.

The structures of the two lactams (VIIIa, b) were proved by consecutive acidic and alkaline treatment, which cleaved the ketal and lactam groups, respectively. The intermediate β -aminoketone (XI) lost ammonia and the unstable cyclohexadiene structure aromatized to @-benzoylpropionic acid (XII) $[m, p. 113-114^\circ; \lambda_{m50}^{ELOH} 206, 241 m_L (e 13200, 12500)]$ in 80% yield. The n.m.r. spectrum of the crude degradation product (XII) showed no proton resonance signals between δ 5.6-6.3 p.p.m., expected for a vinyl ketone derivative (XIII). Thus, the formation of the "isolactam" structure (IX) during ring expansion could be excluded. Because the Beckmann rearrangement proceeds with anti-migration (12), the structure of the two lactams (VIIIa, b) retrospectively verified the <u>anti</u>-stereochemistry^d of both oximes (VIa,b).

Lithium aluminum hydride reduction of the epimer lactams gave rise to the expected $\underline{\text{cis}}$ -aminoketal (Xa, yield 96%)[b.p._{0.001 mm}105-110°; ₆ 2.03 $(10H, m), 3.14 (3H, m), 3.93 (4H, s), 5.64 p.p.m. (2H, b)]$ and trans-aminoketal (Xb) [b.p._{O.O1 mm}98°; δ 2.20 (13H,m), 3.86 (4H,s), 5.50 p.p.m. (2H,b)]. As the closing step of this synthesis, a direct indolization of the cis-aminoketal (Xa) was achieved with sulfuric acid catalysis (13). Both theoretically possible enehydrazine intermediates (12) were apparently present, since the cis-tetracyclic indole (XVIa, yield 70-78%) [hydrochloride: m.p. 266-268°; $\lambda_{\text{max}}^{\text{E}\text{tOH}}$ 226, 283, 291 m_L(₆ 33500, 8400, 7200); free base: 6 2.25 (4H, m), 2.90 (4H, m), 3.45 (2H, m), 5.71 (2H, b), 7.10 (3H, m), 7.48 (1H, m), 7.83 p.p.m.

d) The oxime hydroxyl group is anti to the tertiary bridgehead carbon atom.

(1H, m); R_f 0.47^a] and an indolenine derivative (XVII) $[R_f$ 0.67] were observed. During the acidic cleavage of the ketal group of Xa, partial epimerization occurred and, as a third minor product, the trans-tetracyclic indole (XVIb) $[R_f 0.32]$ was identified. The same indole (XVIb) was obtained by the direct indole-ring closure of the trans-aminoketal (Xb).

An alternative route for the synthesis of the ibogaine model (XVIa) was envisaged. The Fischer indole-ring closure of the cis-enedione-monophenylhydrazone (XVIII, yield 71%) [m.p. 162-163°; $\lambda_{\text{max}}^{\text{RtOH}}$ 277 m_u (e 19400); $\lambda_{\text{max}}^{\text{KBr}}$ 5.85 i (>C=0)] furnished the tetracyclic indole ketone (XIX, yield 80-90%) [m.p. 185-187°; $\lambda_{\text{max}}^{\text{E} \text{CDH}}$ 225, 285, 292 m_{μ} (e 33100, 8900, 7500); $\lambda_{\text{max}}^{\text{KBr}}$ 3.10 (NH), 5.91 $_{\text{H}}$ (x-O)]. During the indolization a complete inversion occurred, both IIa and CIb leading to the same trans-tetracyclic indole derivative (XIX). Although,XIX smoothly underwent oxime formation to XX (yield 85%) [m.p. 224- 225° ; $\lambda_{\text{max}}^{\text{E-OH}}$ 228, 283, 291 m_μ (e 29800, 8300, 7300); $\lambda_{\text{max}}^{\text{KBr}}$ 3.0 μ (OH), no absorption between 5-6 $_{\text{u}}$, its Beckmann rearrangement produced the indolelactam (XXI) $[m \cdot p. 248-250^{\circ}; \lambda_{\max}^{\text{KBr}} 3.10 \text{ (NH)}, 6.05 \mu \text{ (X-A)}]$ in low yield. Because the cis-configuration of the \heartsuit ring could not be retained, there was no further exploration of this approach.

For the total synthesis of ibogaine (I), the Diels-Alder adduct (XXII) $[m.p. 46-48^\circ; \lambda_{\text{max}}^{\text{KBr}} 5.97 \mu \text{ (} \infty=0); 6 0.86 \text{ (}3\text{H}, \text{ t}), 1.38 \text{ (}2\text{H}, \text{ m}), 2.30 \text{ (}3\text{H}, \text{ m}),$ 3.25 (2H. m), 5.75 (2H, m), **6.70** p.p.m. (2H, s)] was selected as a starting material. Selective zinc reduction of XXII provided the cis-enedione (XXIIIa) $[m, p. 71.73^{\circ}, \lambda_{\text{max}}^{\text{KBr}} 5.85 \mu \text{ (&0-0)}; \delta 0.99 \text{ (3H, t)}, 1.54 \text{ (2H, m)}, 2.33 \text{ (3H, m)},$ 2.78 (4H, m), 3.16 (2H, m), 5.75 p.p.m. (2H, m); R_f 0.37^a] in 66% over-all yield, calculated on the trans-1.3-hexadiene. From the endo-cis-formation of XXII, it is believed that the 5-ethyl group of XXIIIa occupies an equatorial position^C. An isomerization of XXIIIa readily gave the trans-epimer (XXIIIb, yield 84%) [m.p. $78-78.6$; $\lambda_{\text{MBK}}^{\text{LB}}$ 5.85 $_{\text{L}}$ (xe=0); $_{\text{6}}$ 0.95 (3H, t), 1.55 (2H, m),

2.28 (3H, m) 2.94 (6H, m) 5.65 p.p.m. (2H, b); R_f 0.42]. The upfield shifted trans C_9 , C_{10} -protons are partially hidden under the signal of the C_2 , C_2 -protons.

* The ring systems of series a and b possess cis and trans configurations, respectively.

The author is indebted to Dr. Charles A. Hetzel and Mrs. Janet T. Watson of this Institute for the nuclear magnetic spectra and vapor-phase chromatography determinations. Satisfactory elemental analyses were obtained for all compounds for which m.p. or b.p. values are cited.

REFERENCES

- 1. The Alkaloids, Academic Press, New York: L. Marion, p. 450 in Vol. II (1952), ed. by R. H. F. Manske and H. L. Holmes; J. E. Saxton, pp. 143-6 in Vol. VII (1960), ed. by R. H. F. Manske.
- 2. 0. Diels and K. Alder, <u>Ber. 62</u>, 2337 (1929).
- 3. R. E. Ireland and J. A. Marshall, <u>J. Org. Chem. 27</u>, 1620 (1962).
- 4. J. E. Cole, W. S. Johnson, P. A. Robins, and J. Walker, <u>J</u>. <u>Chem</u>. <u>Soc</u>. 244 (1962).
- 5. H. B. Henbest, M. Smith, and A. Thomas, <u>J</u>. <u>Chem</u>. <u>Soc</u>. 3293 (1958).
- 6. L. F. Fieser, <u>J</u>. <u>Am</u>. <u>Chem. Soc. 70</u>, 3165 (1948).
- 7. P. Groth and O. Hassel, <u>Proc</u>. <u>Chem</u>. <u>Soc</u>. 218 (1963).
- 8. A. Mossel, C. Romers, and E. Havinga, <u>Tetrahedron Letters</u> 1247 (1963).
- 9. C. Y. Chen and R. J. W. Le Fèvre, Australian J. Chem. 16, 917 (1963).
- 10. L. L. Smith and T. Foell, J. Chromatog. 9, 339 (1962).
- 11. J. C. Craig and A. R. Naik, <u>J. Am. Chem. Soc</u>. 84, 3410 (1962).
- 12. L. G. Donaruma and W. 2. Holdt, in Organic Reactions, Vol. 11, ed. by A. C. Cope et al., p. 4-14, John Wiley + Sons, Inc., New York and London (1960) .
- 13. K. H. Pausacker, J. Chem. Soc. 621 (1950).