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## AN APPROACH TO THE SYNTHESIS OF IBOGAINE

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IN view of the increasing interest in iboga alkaloids I should like to present some of the preliminary steps toward a total synthesis of ibogaine (I) (1). The successful, stereochemically controlled synthesis of the tetracyclic indole (XVIa) suggests a feasible pathway for the synthesis of I and some of its congeners. The <u>cis</u>-fused G'D rings of XVIa were constructed from the <u>cis</u>-enedione (IIa) (2), which, after suitable modifications (IIa  $\longrightarrow$  IIIa  $\longrightarrow$  VIa), was subjected to the Beckmann rearrangement to give VIIIa. The lactam was then reduced to the <u>cis</u>-aminoketal (Xa). The aminoketone derived from Xa underwent indole formation, producing the A-D ring system of XVIa and ibogaine (I).

Parallel transformations of the stereochemically more stable <u>trans</u>enedione (IIb) were carried out, the availability of the <u>trans</u> series (IIIb-VIIIb and Xb) being helpful in determining the configurations during every step of the synthesis. Gas chromatography proved that the separately equilibrated IIa and IIb isomers reached a <u>cis</u>/<u>trans</u> ratio of 1:5.7. In spite of the stereochemical instability of IIa (3,4), it proved to be a useful starting material. The isolated double bond of both epimers (IIa,b) survived all the steps and was found to be particularly helpful in verifying the structures of VIa,b and VIIIa,b (<u>vide infra</u>).

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In reproducing the preparation of the trans-enedione (IIb), as reported by Henbes:, et al. (5), it was proved that the intermolecular chelate (XIV) [m.p. 159-160°; the start and start (>C=0)] was the actual product isolated. The two components of XIV were separated either by thin-layer chromatography (T.L.C.) (R<sub>f</sub> 0.33 and 0.38)<sup>a</sup> or (in 90% yield) by solvolysis. The more puckered cis-dione (IIa) did not chelate with the planar hydroquinone derivative (XV) (6). This difference in behavior offered a means for the isolation of the trans-enedione (IIb) from an equilibrated epimer mixture in 75% over-all yield. The m.p. of IIb was found to be identical (95.5-96.5°) with that reported by Ireland and Marshall (3). The conspicuous differences between the nuclear magnetic resonance (n.m.r.) spectra<sup>b</sup> of the two epimers (IIa, b) established their conformations. The spectrum of IIa indicated non-equivalence for its four C2, C3-protons (sharp peaks at § 2.77 and 2.80 p.p.m.). Contrary to this observation, equivalence of the protons in similar positions of IIb (§ 2.72 p.p.m., s), analogously to cyclohexane-1, 4-dione (7-9), suggested a twisted-boat conformation for its A cing. The axial-equatorial C9, C10-protons of IIa were easily distinguishable ( $\delta$  3.17 p.p.m., m) from the similar but diamagnetically shifted axial-axial protons of IIb (§ 2.59 p.p.m., m).<sup>C</sup>

a) The T.L.C. systems were : Al<sub>2</sub>O<sub>3</sub>-G[ethyl acetate - n-hexane (2:3)] for XIV; neutral silica gel - starch (10) [ethyl acetate - n-hexane (2:3) for IIa,b to VIa,b and XXIIIa,b; Al<sub>2</sub>O<sub>3</sub>-G[chloroform] for VIIIa,b; Al<sub>2</sub>O<sub>3</sub>-G[chloroform - cyclohexane - diethylamine (7:2:1)] for XVIa,b and XVII.

b) Measured in deuteriochloroform at 60 Mc on a Varian, Model A-60, spectrometer and expressed as p.p.m. shift ( $\delta$ ) downfield from tetramethylsilane.

c) Details of the conformational analyses will be published elsewhere.



XIII

XVIa,b

The ring systems of series a and b possess cis and trans configurations, \* respectively.

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The cis-dione (IIa) failed to produce a satisfactory yield of the <u>cis</u>-monexime (Va) [ $\lambda_{max}^{KBr}$  3.15 (OH), 5.85  $\mu$  (>C=0); R<sub>f</sub> 0.17<sup>a</sup>] and, furthermore, Va could not be rearranged to the corresponding lactam. To achieve a higher degree of stereostability before the oxime function was introduced, the monoketalization of IIa was studied (3,4). The ketalization, which was followed by quantitative T.L.C., led to a mixture of the cis-monoketal (IIIa, yield ca. 27%) [m.p. 62-64°; λ<sup>KBr</sup><sub>max</sub> 5.85 μ; δ 2.30 (8H,m), 3.09 (2H,m), 4.0 (4H,d), 5.52 p.p.m. (2H,b); Rf 0.43] and the cis-bisketal (IVa, yield 10-15%) [m.p. 116-117° (4); § 1.77 (4H,m), 2.13 (6H,b), 3.86 (8H,s), 5.54 p.p.m. (2H,b);  $R_{f}$  0.53], as well as to the trans-monoketal (IIIb, yield ca. 22%) [m.p. 52-54° (4); λ<sup>KB::</sup> 5.85 μ; δ 2.03 (8H,m), 2.56 (2H,m), 3.98 (4H,b), 5.53 p.p.m. (2H,b); R<sub>f</sub> 0.48] and the <u>trans</u>-bisketal (IVb, yield 15-30%) [m.p. 97.5-98° (3); § 1.71 (4H,s), 2.01 (6H,s), 3.86 (8H,b), 5.52 p.p.m. (2H,b); R<sub>f</sub> 0.55]. Some starting material (IIa, ca. 5%) ( $R_f$  0.24) and its epimer (IIb, ca. 10%) ( $R_f$  0.35) were also detected. Thus, the T.L.C. and n.m.r. data revealed the concomitant epimerization and two-step ketalization of IIa. It was also proved that the cis-monoketal (IIIa) did not equilibrate with the trans-epimer (IIIb) during isolation from a Florisil column, as reported by others (4).

The two monoketals (IIIa,b) retained their configurations during oximation and gave rise to the <u>cis-anti</u>-oximeketal (VIa, yield 27% calculated on IIa) [m.f. 178-179°;  $\lambda_{max}^{KBr}$  3.15 (OH), 5.98 (>C=N-), 6.03  $\mu$  (>C=C<); R<sub>f</sub> 0.46] and <u>trans-anti</u>-oximeketal (VIb, yield 22% calculated on IIa) [m.p. 166-167°;  $\lambda_{max}^{KBr}$  3.15 (OH), 5.98 (>C=N-), 6.03  $\mu$  (>C=C<); R<sub>f</sub> 0.54]. The <u>anti</u>-oxime structure of VIa,b was verified by degradation (<u>vide infra</u>). On tosylation of the epimer ketaloximes (VIa,b) in warm pyridine, they exhibited a remarkable difference. While the <u>trans</u>-epimer (VIb) furnished the expected <u>trans</u>tosyloxime ketal (VIIb) (m.p. 131-132°) in almost quantitative yield, the <u>cis</u>-epime: (VIa) spontaneously rearranged to the desired <u>cis</u>-lactamketal (VIIIa, yield 89%) [m.p. 211.5-212°;  $\lambda_{max}^{KBr}$  3.15 (NH), 6.03  $\mu$  (>C=O); R<sub>f</sub> 0.34<sup>a</sup>].

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It is believed that the coplanarity of the participating centers of the intermediate <u>cis</u>-tosyloxime (VIIa) facilitate the Beckmann rearrangement. The <u>trans</u>-tosyloxime (VIIb) was ring expanded to the trans-lactamketal (VIIIb) [m.p. 200°;  $\lambda_{max}^{KBr}$  3.15 (NH), 6.0  $\mu$  (>C=O); R<sub>f</sub> 0.20] on a basic aluminum oxide column (11) in 78% yield.

The structures of the two lactams (VIIIa,b) were proved by consecutive acidic and alkaline treatment, which cleaved the ketal and lactam groups, respectively. The intermediate  $\beta$ -aminoketone (XI) lost ammonia and the unstable cyclohexadiene structure aromatized to  $\beta$ -benzoylpropionic acid (XII) [m.p. 113-114°;  $\lambda_{max}^{EtOH}$  206,241 m<sub>µ</sub> ( $_{c}$  13200, 12500)] in 80% yield. The n.m.r. spectrum of the crude degradation product (XII) showed no proton resonance signals between  $\delta$  5.6-6.3 p.p.m., expected for a vinyl ketone derivative (XIII). Thus, the formation of the "isolactam" structure (IX) during ring expansion could be excluded. Because the Beckmann rearrangement proceeds with <u>anti</u>-migration (12), the structure of the two lactams (VIIIa,b) retrospectively verified the <u>anti</u>-stereochemistry<sup>d</sup> of both oximes (VIa,b).

Lithium aluminum hydride reduction of the epimer lactams gave rise to the expected <u>cis</u>-aminoketal (Xa, yield 96%)[b.p.<sub>0.001 mm</sub>105-110°;  $\delta$  2.03 (10H, m), 3.14 (3H,m), 3.93 (4H,s), 5.64 p.p.m. (2H,b)] and <u>trans</u>-aminoketal (Xb) [b.p.<sub>0.01 mm</sub>98°;  $\delta$  2.20 (13H,m), 3.86 (4H,s), 5.50 p.p.m. (2H,b)]. As the closing step of this synthesis, a direct indolization of the <u>cis</u>-aminoketal (Xa) was achieved with sulfuric acid catalysis (13). Both theoretically possible enchydrazine intermediates (12) were apparently present, since the <u>cis</u>-tetracyclic indole (XVIa, yield 70-78%) [hydrochloride: m.p. 266-268°;  $\lambda_{max}^{EtOH}$  226, 283, 291 m<sub>L</sub> ( $\epsilon$  33500, 8400, 7200); free base:  $\delta$  2.25 (4H, m), 2.90 (4H, m), 3.45 (2H, m), 5.71 (2H, b), 7.10 (3H, m), 7.48 (1H, m), 7.83 p.p.m.

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d) The oxime hydroxyl group is <u>anti</u> to the tertiary bridgehead carbon atom.

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(1H, m);  $R_f 0.47^a$ ] and an indolenine derivative (XVII)  $[R_f 0.67]$  were observed. During the acidic cleavage of the ketal group of Xa, partial epimerization occurred and, as a third minor product, the <u>trans</u>-tetracyclic indole (XVIb)  $[R_f 0.32]$  was identified. The same indole (XVIb) was obtained by the direct indole-ring closure of the trans-aminoketal (Xb).

An alternative route for the synthesis of the ibogaine model (XVIa) was envisaged. The Fischer indole-ring closure of the <u>cis</u>-enedione-monophenylhydrazone (XVIII, yield 71%) [m.p. 162-163°;  $\lambda_{max}^{EtOH} 277 m_{\mu}$  (e 19400);  $\lambda_{max}^{KBr} 5.85 \mu$ (>C=0)] furnished the tetracyclic indole ketone (XIX, yield 80-90%) [m.p. 185-187°;  $\lambda_{max}^{EtOH} 225$ , 285, 292 m<sub> $\mu$ </sub> (e 33100, 8900, 7500);  $\lambda_{max}^{KBr} 3.10$  (NH), 5.91  $\mu$  (>C=0)]. During the indolization a complete inversion occurred, both IIa and IIb leading to the same <u>trans</u>-tetracyclic indole derivative (XIX). Although, XIX smoothly underwent oxime formation to XX (yield 85%) [m.p. 224-225°;  $\lambda_{mix}^{E-OH} 228$ , 283, 291 m $_{\mu}$  (e 29800, 8300, 7300);  $\lambda_{max}^{KBr} 3.0 \mu$  (OH), no absorption between 5-6  $\mu$ ], its Beckmann rearrangement produced the indolelactam (XXI) [m.p. 248-250°;  $\lambda_{max}^{KBr} 3.10$  (NH), 6.05  $\mu$  (>C=0)] in low yield. Because the <u>cis</u>-configuration of the C/D ring could not be retained, there was no further exploration of this approach.

For the total synthesis of ibogaine (I), the Diels-Alder adduct (XXII) [m.p. 46-48°;  $\lambda_{\text{max}}^{\text{KBr}}$  5.97  $\mu$  (>C=0);  $\delta$  0.86 (3H, t), 1.38 (2H, m), 2.30 (3H, m), 3.25 (2H m), 5.75 (2H, m), 6.70 p.p.m. (2H, s)] was selected as a starting material. Selective zinc reduction of XXII provided the <u>cis</u>-enedione (XXIIIa) [m.p. 71-73°,  $\lambda_{\text{max}}^{\text{KBr}}$  5.85  $\mu$  (>C=0);  $\delta$  0.99 (3H, t), 1.54 (2H, m), 2.33 (3H, m), 2.78 (4H, m), 3.16 (2H, m), 5.75 p.p.m. (2H, m);  $R_{\rm f}$  0.37<sup>a</sup>] in 66% over-all yield, calculated on the <u>trans</u>-1.3-hexadiene. From the <u>endo-cis</u>-formation of XXII, it is believed that the 5-ethyl group of XXIIIa occupies an equatorial position<sup>c</sup>. An isomerization of XXIIIa readily gave the <u>trans</u>-epimer (XXIIIb, yield 84%) [m.p. 78-78.6;  $\lambda_{\rm KBI}^{\text{KBI}}$  5.85  $\mu$  (>C=0);  $\delta$  0.95 (3H, t), 1.55 (2H, m),

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2.28 (3H, m) 2.94 (6H, m) 5.65 p.p.m. (2H, b);  $R_f$  0.42]. The upfield shifted trans C<sub>9</sub>, C<sub>10</sub>-protons are partially hidden under the signal of the C<sub>2</sub>, C<sub>3</sub>-protons.



\* The ring systems of series <u>a</u> and <u>b</u> possess <u>cis</u> and <u>trans</u> configurations, respectively.

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## REFERENCES

- <u>The Alkaloids</u>, Academic Press, New York: L. Marion, p. 450 in Vol. II (1952), ed. by R. H. F. Manske and H. L. Holmes; J. E. Saxton, pp. 143-6 in Vol. VII (1960), ed. by R. H. F. Manske.
- 2. O. Diels and K. Alder, Ber. 62, 2337 (1929).
- 3. R. E. Ireland and J. A. Marshall, J. Org. Chem. 27, 1620 (1962).
- J. E. Cole, W. S. Johnson, P. A. Robins, and J. Walker, <u>J. Chem. Soc</u>. 244 (1962).
- 5. H. B. Henbest, M. Smith, and A. Thomas, J. Chem. Soc. 3293 (1958).
- 6. L. F. Fieser, J. Am. Chem. Soc. 70, 3165 (1948).
- 7. P. Groth and O. Hassel, Proc. Chem. Soc. 218 (1963).
- 8. A. Mossel, C. Romers, and E. Havinga, Tetrahedron Letters 1247 (1963).
- 9. C. Y. Chen and R. J. W. Le Fèvre, Australian J. Chem. 16, 917 (1963).
- 10. L. L. Smith and T. Foell, J. Chromatog. 9, 339 (1962).
- 11. J. C. Craig and A. R. Naik, J. Am. Chem. Soc. 84, 3410 (1962).
- L. G. Donaruma and W. Z. Holdt, in <u>Organic Reactions</u>, Vol. 11, ed. by A. C. Cope <u>et al.</u>, p. 4-14, John Wiley 4 Sons, Inc., New York and London (1960).
- 13. K. H. Pausacker, J. Chem. Soc. 621 (1950).